# FORMATION OF FOAMED STRUCTURE IN ALUMINUM AND IRON MELTS AT HIGH-RATE TENSION

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### Problem statement

Evolution of metal surface after powerful 100 fs laser irradiation (MD simulations)



[Roth J., Sonntag S., Karlin J., and Trebin H.-R. // InSiDe (Innovatives Supercomputing in Deutschland) (2012)]

Irradiation of tantalum film by powerful ultra-short laser pulse (MD simulations)



[Inogamov N.A., et al. // Contrib. Plasma Phys. (2013)]

### High-rate tension of Al melt

LAMMPS [Plimpton S. // J. Comp. Phys. (1995) http://lammps.sandia.gov] OVITO [Stukowski A. // Modell. Simul. Mater. Sci. Eng. (2010) http://www.ovito.org] "Construct surface mesh" algorithm [Stukowski A. // JOM (2014)]

Initial temperature is 2500 K to ensure melting

Uniform tension at constant temperature of 1100 K

Strain rates (true) are 3, 10, 30 and 100/ns

Degree of deformation is from 2.5 to 10





laser solid [Ashitkov S.I., Agranat M.B., Kanel G.I., Komarov P.S., Fortov V.E. // JETP Lett. (2010)]
laser melt [Agranat M.B., Anisimov S.I., Ashitkov S.I.,

Zhakhovskii V.V., Inogamov N.A., Komarov P.S., Ovchinnikov A.V., Fortov V.E., Khokhlov V.A., Shepelev V.V. // JETP Lett. (2010)]

MD [Mayer A.E., Mayer P.N. // Comp. Mat. Sci. (2016)] 3



#### Late stages of high-rate tension of Al melt: cavities and foamed melt



Strain rate is 30/ns, temperature is 1100K

#### Late stages of high-rate tension of Al melt: breaking of walls and decay on jets

110 ps 130 ps 3.3 3.9 170 ps 150 ps 4.5 5.1  $V / V_0 = 164$ Ż, y z y

Strain rate is 30/ns, temperature is 1100K

#### Late stages of high-rate tension of Al melt: breaking of jets on droplets



Strain rate is 30/ns, temperature is 1100K

**Evolution of pores in Al melt at tension** 

Estimations: 
$$R_p = 3 \frac{(V - V_s)}{S}$$
  $N_p = \frac{1}{36\pi} \frac{S^3}{(V - V_s)^2}$ 

S is surface area

V is total volume  $V_s$  is liquid volume



[Mayer P.N., Mayer A.E. // J. Appl. Phys. (2016)]

#### **Characteristics of droplets**



Correlation between numbers of pores and droplets

[Mayer P.N., Mayer A.E. // J. Appl. Phys. (2016)]

#### Pressure at the stage of foamed Al melt



temperature is 1100K

=> investigation of the void size distribution is required

#### Work spent on deformation of MD system



[Mayer P.N., Mayer A.E. // J. Appl. Phys. (2016)]

## High-rate tension of Fe melt

LAMMPS [Plimpton S. // J. Comp. Phys. (1995) http://lammps.sandia.gov] OVITO [Stukowski A. // Modell. Simul. Mater. Sci. Eng. (2010) http://www.ovito.org] "Construct surface mesh" algorithm [Stukowski A. // JOM (2014)]

Initial temperature is 3000 K to ensure melting

Uniform tension at constant temperature of 2000 K

Strain rates (true) 10, 30 and 100/ns

Maximal degree of deformation is about 11



Interatomic potential for Fe [Zhou, X.W., Johnson, R.A., Wadley, H.N.G. // Phys. Rev. B. (2004)]

Tensile strength of Fe melt at the strain rate of (0.4-0.9)/ns and temperature of 3000 K:

0.5-1.3 GPa according to the experimental data [Struleva, E.V., Ashitkov, S.I., Komarov, P.S., Khishchenko, K.V., Agranat, M.B., // J. Phys.: Conf. Ser. (2016)]
2 GPa according to MD simulations [Mayer A.E., Mayer P.N. // J. Appl. Phys. (2015)] with this potential

#### Nucleation, growth and consolidation of cavities in foamed Fe melt temperature is 2000 K; strain rate is 30/ns



#### **Destruction of foamed melt structure of jets and decay of jets on droplets** Fe melt, temperature is 2000 K; strain rate is 30/ns



#### Nucleation, growth and consolidation of cavities in foamed Fe melt temperature is 2000 K; strain rate is 100/ns

**8 ps 4 ps**  $\varepsilon = 0.4$  $\varepsilon = 0.8$ V/V0 = 1.5V/V0 = 2.2Z Y end c\_eng 12 ps 20 ps  $\varepsilon = 1.2$  $\varepsilon = 2$ V/V0 = 3.3*V*/*V*0 = 7.4 c eng

#### **Destruction of foamed melt structure of jets and decay of jets on droplets** Fe melt, temperature is 2000 K; strain rate is 100/ns



#### Pressure at the stage of foamed Fe melt

temperature is 2000K



### **Evolution of pores in Fe melt at tension**

temperature is 2000K



10

Time (ps)

100

1000

17

for Fe melt than that for Al melt

## Solidification of foamed Al melt

LAMMPS [Plimpton S. // J. Comp. Phys. (1995) http://lammps.sandia.gov] OVITO [Stukowski A. // Modell. Simul. Mater. Sci. Eng. (2010) http://www.ovito.org] "Construct surface mesh" algorithm [Stukowski A. // JOM (2014)]

Initial temperature is 2500 K to ensure melting

Cooling to 1100 K during 10 ps at zero pressure

Uniform tension with true strain rate of 30/ns during 50 ps: final degree of deformation is 1.5 volume increases in 4.5 times

Two stages of cooling:

- 1. Cooling from 1100 to 500 K during 200 ps  $\left(\frac{dT}{dt} = 3 \cdot 10^{12} \text{ K}^{-1}\right)$
- 2. Cooling from 500 to 300 K during 200 ps

Interatomic potential for Al [Zope R.R. and Mishin Y. // Phys. Rev. B. 68, 024102 (2003)]

#### 4 000 000 atoms



$$\left(\frac{dT}{dt} = 10^{12} \text{ K}^{-1}\right)$$

18

#### **Evolution of foamed structure in Al during the cooling of melt at constant volume**



Growth of large cavities and collapse of small cavities instead of constant volume

### **Stabile form of cavities in Al at low temperatures**



Stabile form argue that we obtain a solidified structure

Crystalline lattice is not detected

Week elastic properties are observed during subsequent uniaxial tension 20

### **Conclusions**

- 1. Complete fracture of melt includes the following stages: (i) growth of large and collapse of small pores; (ii) destruction of walls between pores with formation of jets; (iii) fragmentation of jets on droplets.
- The foamed melt retains till the void volume fraction exceeds 0.9 at least. 2.
- 3. Pressure remains negative in the foamed melt at tension. The work required for complete fracture exceeds several times the work on reaching the cavitations point.
- 4. The foamed structure evolution is controlled by surface tension.
- 5. Analytical estimations for the time evolution of the mean radius, pressure in the system and the work on melt tension are proposed.
- 6. Simulation of the foamed melt cooling down to room temperature shows a solidified foamed metal structure formation, which persists over time.

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